

## Chemistry Worksheets Class 10 on Chapter 2 Acids, Bases and Salts - Set 2

**Q1.** Fresh milk has a pH of 6. When milk changes into curd, the pH value will:

- a.) become 7
- b.) become less than 6
- c.) become more than 7
- d.) remains unchanged

**Q2.** Which of the following is medicine for indigestion?

- a.) sodium hydroxide
- b.) manganese hydroxide
- c.) magnesium hydroxide
- d.) potassium hydroxide

**Q3.** How many number of water of crystallisation is present in copper sulphate crystals?

- a.) 3
- b.) 5
- c.) 7
- d.) None

**Q4.** The indicators which turn red in acid solution are:

- a.) turmeric and litmus
- b.) phenolphthalein and methyl orange
- c.) litmus and methyl orange
- d.) phenolphthalein and litmus

**Q5.** The salt which will give an acidic solution on dissolving in water is:

- a.) KCl
- b.)  $\text{NH}_4\text{Cl}$
- c.)  $\text{Na}_2\text{CO}_3$
- d.)  $\text{CH}_3\text{COONa}$

**Q6.** What are the observations when quick lime is added to water?

**Q7.** Which indicator gives pink colour in the basic solution?

# Chapter 15 Acids Bases Section 2 Answers

**Molly M. Bloomfield, Lawrence J.  
Stephens**



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**CliffsNotes AP Chemistry** Angela Woodward Spangenberg, 2016-01-12 Test prep for the AP Chemistry exam with 100% brand new content that reflects recent exam changes Addressing the major overhaul that the College Board recently made to the AP Chemistry exam this AP Chemistry test prep guide includes completely brand new content tailored to the exam administered every May Features of the guide include review sections of the six big ideas that the new exam focuses on Fundamental building blocks Molecules and interactions Chemical reactions Reaction rates Thermodynamics Chemical equilibrium Every section includes review questions and answers Also included in the guide are two full length practice tests as well as a math review section and sixteen discrete laboratory exercises to prepare AP Chemistry students for the required laboratory experiments section on the exam

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