

## REVIEW OF BASIC CHEMISTRY ANSWER KEY

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1. Name the following elements. Spelling counts:

H *hydrogen*  
S *sulphur*  
Cl *chlorine*  
C *carbon*  
N *nitrogen*  
Na *sodium*  
Pb *lead*  
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*  
phosphorus *P*  
fluorine *F*  
magnesium *Mg*  
potassium *K*  
calcium *Ca*  
zinc *Zn*  
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79**                      copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5**              calcium (Ca) **40.1**

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5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
<b>Lead</b>	<b>82</b>	<b>207</b>	<b>82</b>	125	<b>82</b>
<b>Argon</b>	<b>18</b>	<b>56</b>	18	40	<b>18</b>
Barium	<b>56</b>	137	<b>56</b>	<b>61</b>	<b>56</b>
<b>Tin</b>	<b>50</b>	112	<b>50</b>	<b>62</b>	50

6. Match the unit with its description:

F 1. amount of substance  
C 2. density  
B 3. mass  
D 4. molar mass  
A 5. energy  
E 6. volume

A. joule  
B. g  
C. g/mL  
D. g/mole  
E. L  
F. mole

# Chapter 12 Basic Review Worksheet Chemistry Answers

**Ying-Ying Zheng**



## Chapter 12 Basic Review Worksheet Chemistry Answers:

*Science Insights*, 1999      **Books in Print Supplement**, 2002      **Linne & Ringsrud's Clinical Laboratory Science - E-Book** Mary Louise Turgeon, 2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts      *Biology* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004      **Characterizing Pedagogical Flow** W.H. Schmidt, 2005-12-17 Characterizing Pedagogical Flow presents conclusions from a multi disciplinary multi national research project blending quantitative and qualitative approaches through a discourse methodology The work produced portraits of mathematics and science education that were dramatically different for each of the countries involved France Japan Norway Spain Switzerland and the United States To explain these differences it is proposed that the interaction of curriculum and pedagogy is culturally unique and yields classroom learning experiences that are qualitatively different from country to country This idea has profound implications for how international education research is interpreted      **Holt Chemistry** Ralph Thomas Myers, 2004

*Curriculum Review*, 1986      *Programmed Learning and Individually Paced Instruction* Carl H. Hendershot, 1973      Te HS&T 2007 Shrt Crs M Holt Rinehart & Winston, 2007      Glencoe Science, 2002      R.R. Bowker's Software for Schools, 1987      **Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03      *A Level Chemistry Multiple Choice Questions and Answers (MCQs)* Arshad Iqbal, 2020-04-10 Previously published as A Level Chemistry MCQs Multiple Choice Questions and Answers Quiz Tests with Answer Keys by Arshad Iqbal A Level Chemistry Multiple Choice Questions and Answers MCQs A Level Chemistry quizzes practice tests with answer key provides mock tests for competitive exams to solve 1745 MCQs A Level Chemistry MCQs helps with theoretical conceptual and analytical study for self assessment career tests This book can help to learn and practice A Level Chemistry quizzes as a quick study guide for placement test preparation A level Chemistry Multiple Choice Questions and Answers MCQs is a revision guide with a collection of trivia quiz questions and answers on topics Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements to enhance teaching and learning A level Chemistry Quiz Questions and Answers also covers the syllabus of many competitive papers for admission exams of different universities from chemistry textbooks on chapters Alcohols and Esters Multiple Choice Questions 27 MCQs Atomic Structure and Theory Multiple Choice Questions 37 MCQs Benzene Chemical

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