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Radiologic Science for Technologists Stewart C. Bushong,1993 This text aims to bring students up to date with radiologic science Now in its fifth edition it includes new chapters such as image contrast and fast imaging techniques of MRI duplex technology of diagnostic ultrasound and quality assurance in x ray imaging *Byte* ,1996-05 **InfoWorld** ,1991 **Business China** ,1987

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 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$
 $2\text{C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SnO}_4 + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3$
 $3\text{NF}_3 + \text{SnF}_2 \rightarrow \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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3 4 5 1 $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
3 3 $\text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{H}_2\text{O} + 3\text{H}_2\text{SO}_4 \rightarrow 3\text{NaBr} + 3\text{SnO}_2 + 3\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$ using the following equation
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu} + \text{Zn} + \text{CuCl}_x$
 $2\text{ZnCl}_x + 2\text{Cu}$ how many moles of $\text{Ce} + \text{ZnCl}_2 + \text{ZnCl}_x + 2$ will be produced from 23.0 g of $\text{Ce} + \text{Zn}$ assuming $\text{Ce} + \text{CuCl}_2 + \text{CuCl}_x + 2$ is available in excess

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