

Chemistry I (H) - Unit 02 (Pt. II - Nuclear Chemistry) Study Guide

- 1) Define, compare, and contrast radiation (pg. 122) to radioactivity (pg. 122).
- 2) Define AND give an example of a radioisotope (pg. 861).
- 3) Define nuclear reactions.
 - Define nuclear equations (pg. 123/869) -
 - Example:
- 4) Define and provide examples of the following forms of radioactive decay:
 - alpha decay (pg. 867) -
 - alpha particle (pg. 123/862) -
 - Example Nuclear Equation:
 - beta decay (pg. 867) -
 - beta particle (pg. 123/863) -
 - Example Nuclear Equation:
 - gamma radiation (pg. 124) -
 - gamma ray (pg. 124/863) -
 - Example Nuclear Equation:
 - Study the Table 24.2 from pg. 861 that we drew in our notes about the properties of decay (penetrating power, mass, etc.)
- 5) Define AND give an example of transmutation (pg. 865).
- 6) What do we mean by the words emit or emanate? Give an example.
- 7) Compare a *stable nucleus* to that of an *unstable nucleus*.

Section 25 Nuclear Chemistry Study Guide Answers

Susan McMurry



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Nuclear Science Abstracts, 1971

Study Guide General Chemistry, Second Edition, Becker/Wentworth Becker, Benjamin L. Carroll, 1980

Study Guide and Solutions Manual for Organic Chemistry Susan McMurry, 1996 This revision of the best selling organic chemistry textbook today has been fully updated and revised to offer more applications a completely new chapter and dozens of new problems and examples McMurry's text is currently in use at hundreds of colleges and universities throughout the United States and Canada and is an international bestseller from the United Kingdom to the Pacific Rim In this edition McMurry continues to do what he does best focus on the important material of the course and explain it in a concise clear way

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