

6. In the reaction of iron with copper sulphate solution:



Which option in the given table correctly represents the substance oxidised and the reducing agent?

OPTION	Substance Oxidized	Reducing Agent
A	Fe	Fe
B	Fe	FeSO ₄
C	Cu	Fe
D	CuSO ₄	Fe

7. The chemical reaction between copper and oxygen can be categorized as:

- A. Displacement reaction
- B. Decomposition reaction
- C. Combination reaction
- D. Double displacement reaction

8. Which of the given options correctly represents the Parent acid and base of Calcium Carbonate?

OPTION	PARENT ACID	PARENT BASE
A	HCl	NaOH
B	H ₂ CO ₃	Ca(OH) ₂
C	H ₃ PO ₃	CaSO ₄
D	H ₂ SO ₄	CaSO ₄

9. How will you protect yourself from the heat generated while diluting a concentrated acid?

- A. By adding acid to water with constant stirring.
- B. By adding water to acid with constant stirring.
- C. By adding water to acid followed by base.
- D. By adding base to acid with constant stirring.

10. Why is it important to balance a skeletal chemical equation?

- A. To verify law of conservation of energy.
- B. To verify the law of constant proportion.
- C. To verify the law of conservation of mass.
- D. To verify the law of conservation of momentum.

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